

VPG (4) — Chem
(14) Gr. A

2016-18

Time : 3 hours

Full Marks : 70

*Candidates are required to give their answers in
their own words as far as practicable.*

The figures in the margin indicate full marks.

*Answer **five** questions in which*

Q. No. 1 is compulsory.

1. Answer **all** questions : 2×7 = 14
- (a) What are dinitrogen complexes ? Give an example.
- (b) Why water can't be used as the solvent for infrared spectroscopy ?
- (c) What are different term-symbols of a d^2 system ?

- (d) Define Racah Parameter:
- (e) Give the name of four Massbauer nucleides.
- (f) NO_2 Show esr spectrum but CO_2 doesnot. Explain.
- (g) What is g-values in esr spectra ?
2. (a) How IR spectroscopy is useful in the elucidation of structures of metal nitrosyls. 7
- (b) Explain, with suitable examples, where NO acts as terminal linear, bent and bridging ligands in different complexes. 7
3. (a) Explain the orbital and spin selection rules in transition metal complexes. What are forbidden transitions. 6+2 = 8
- (b) Discuss the factors determining the width of the transition metal complex electronic spectra. 6
4. (a) Explain the splitting of Russel-Saunders states S, P, D and F in octahedral and tetrahedral fields by Mulliken Symbol. 8

- (b) Draw Orgel diagram for d^2 system and predict the different transition as per Orgel diagram. 6
5. (a) Outline the basic Principles of esr spectrum. 6
- (b) Predict the number of spectral lines in esr spectrum of :
- (i) $[\text{Co}(\text{H}_2\text{O})_6]^{2+}$
- (ii) $[\text{Cr}(\text{H}_2\text{O})_6]^{2+}$
- Indicate how zero field splitting and Krammer's degeneracy applies in these compounds. http://www.vbuonline.com 8
6. What is Massbauer spectroscopy ? Describe its application indetermining structure and bonding. 14
7. (a) Explain the structure of B_2H_6 by NMR Spectroscopy. 6
- (b) Explain the use of ^{13}C and ^{31}P NMR in the determination of structure of Inorganic molecules. 8

8. (a) What are the characteristics of Massbauer nucleides ? 6
- (b) Explain the structure of Prussian blue on the MB spectroscopy. 8
9. Write short notes on any two of the following : $7 \times 2 = 14$
- (a) CHEMICAL shift in NMR spectroscopy
- (b) Magnetically ordered compounds and their use.
- (c) Zero-field splitting and Krammer's degeneracy
- (d) Tanabe-Sugano diagram



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