

VPG (4) — Chem  
(13) Gr. C

**2016-18**

Time : 3 hours

Full Marks : 70

Candidates are required to give their answers in their own words as far as practicable.

The figures in the margin indicate full marks.

Answer any **five** questions.

1. (a) Define hydrogen molecule ion.  
(b) Establish the evaluation of  $\psi$  and  $\psi^2$ .  
5+9 = 14
2. State and explain the following : 7×2 = 14  
(a) Born Oppenheimer Equation  
(b) Roothann's Equation
3. Explain the Application of HMO theory to Benzene. 14

CB – 11/2

( Turn over )

4. (a) Discuss Heteronuclear Conjugated System of Pyrrole.

- (b) Derive Electronic energy of molecule.

9+5 = 14

5. What is scattering length ? Explain Low energy scattering theory. 4+10 = 14

6. Explain the combination of one 2S and three 2P orbital. http://www.vbuonline.com 14

7. Define Levinson's Theorem. Write the application of scattering theory in square well Potential. 14

8. Write short notes on any **two** of the following : 7×2 = 14

- (a) Breit-wigner Formula
- (b) Levinson's Theorem
- (c) Coulomb wave function

9. Explain the Density function theory and discuss its application. 7+ 7 = 14



CB – 11/2 (300)

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